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A kinetic study of poly(N-isopropylacrylamide) in the presence of DNA under hydrothermal conditions. The effect of hydrothermal treatment on the thermally reversible phase transition of poly(N-isopropylacrylamide), or PNIPAAm, was studied by using UV-Vis spectroscopy and differential scanning calorimetry (DSC). In the absence of DNA (samples PNIPAAm/DNA), a uniform increase of the critical temperature of phase transition (T_c) on increasing the hydrothermal treatment time was observed, from 45.3 to 61.3 degrees C, in samples for which the weight concentration of polymer was 5.6% (w/w). In the presence of an ionic surfactant, in the hydrothermal reaction mixture (samples PNIPAAm/DNA/surfactant) the T_c value remained stable. The effect of the ionic surfactant on PNIPAAm/DNA/surfactant systems, in terms of the T_c value, was similar to the effect of increasing the hydrothermal reaction time. We propose that in our systems, the polymer chains migrate into the bulk solution during the hydrothermal reaction of the PNIPAAm, resulting in a modification of the polymer structure that can be measured by UV-Vis spectroscopy. This migration is presumably promoted by the presence of the ionic surfactant, which, due to its hydrophilic character, favors the formation of the globular micelle structure and of the PNIPAAm/surfactant complex in solution. 0 * j * * 3

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